em	Chemical	Species	Carcinogenic Risk	Characteristics	Consequence	IARC Group
	Sulphur dioxide	Criteria air pollutants	-	Sulphur dioxide is a highly toxic, colorless with a choking or suffocating odour, nonflammable gas. It is used as a pharmaceutical aid and antioxidant. It is also an environmental air pollutant	Acute exposure of sulphur dioxide would resulting in irritation and lacrimation to eyes. Chronic exposure of Sulphur dioxide can result in an altered sense of smell (including increased tolerance to low levels of sulphur dioxide), increased susceptibility to respiratory infections, symptoms of chronic bronchitis, and accelerated decline in pulmonary function.	Group 3
	Nitrogen dioxide	Criteria air pollutants	-	Nitrogen dioxide is part of a group of gaseous air pollutants produced as a result of road traffic and other fossil fuel combustion processes.	Acute exposure of nitrogen dioxide could result in eye, throat and lung irritation and triggers asthma in asthmatics. Chronic exposure of Nitrogen dioxide may decrease lung fuction and increase the risk of respiratory symptoms such as acute bronchitis and cough and phlegm, particularly in children.	-
	Carbon monoxide	Criteria air pollutants	-	Carbon Monoxide is an odorless, tasteless, poisonous gas, CO, that results from the incomplete combustion of carbon. Inhalation causes central nervous system damage and asphyxiation. It is found in bothoutdoor and indoor air.	Acute effects of carbon monoxide would have symptoms in order of increasing severity of CO poisoning are headache, dizziness on exertion, fatigue, palpitations, nausea, vomiting, difficulty breathing on exertion, mental confusion, rapid heartbeat, visual disturbance, and muscle twitch. Chronic exposure to low levels of CO can lead to a cluster of symptoms resembling the flu: headache, fatigue, muscle aches, nausea, vomiting, and a change in sensitivity to light, odor, and taste.	-
	Hydrogen sulphide	Inorganic compound	-	Hydrogen sulphide is a flammable, poisonous gas with a characteristic odor of rotten eggs. Hydrogen sulphide (H2S) occurs naturally in crude petroleum, natural gas, volcanic gases, and hot springs.	Acute and chronic exposure of Hydrogen sulphide may result in may result in effects on the respiratory, neurological, ocular and cardiovascular systems.	-
	Ammonia	Inorganic compound	-	Ammonia is a colorless inorganic compound of nitrogen and hydrogen with the formula NH3, usually in gaseous form with a characteristic pungent odor	Acute exposure of Ammonia to humans can initially cause upper respiratory tract irritation and chronic inhalation is associated with increased cough, phlegm, wheeze and asthma.	-
	Dimethyl sulphide	Organosulfur compounds	-	Dimethyl sulfide is a clear colorless to straw colored liquid with a disagreeable odor. Flash point less than 0 °F. Less dense than water and slightly soluble in water. Vapors are heavier than air.	Acute exposure of humans to the vapors of dimethyl sulfate may cause severe inflammation and necrosis of the eyes, mouth, and respiratory tract; severe damage to the lungs may result. No information is available on the chronic effects of dimethyl sulfate in humans.	-
	Diethyl sulphide	Organosulfur compounds	-	Diethyl sulfide is a colorless oily liquid with a garlic-like odor. Less dense than water. Vapors heavier than air. May irritate skin and eyes.	May cause toxic effects if inhaled or absorbed through skin. Inhalation or contact with material may irritate or burn skin and eyes. Fire will produce irritating, corrosive and/or toxic gases. Vapors may cause dizziness or suffocation. Runoff from fire control or dilution water may cause pollution.	-
	Acetone	Ketone	-	Acetone is a colorless, volatile, flammable organic solvent. Acetone occurs naturally in plants, trees, forest fires, vehicle exhaust and as a breakdown product of animal fat metabolism. This agent may be normally present in very small quantities in urine and blood; larger amounts may be found in the urine and blood of diabetics. Acetone is toxic in high doses.	Acute exposure to acetone can cause eye irritation, dryness of the mouth and throat, nausea, vomiting, headache, sleepiness, dizziness, light-headedness, weakness, incoordination, loss of energy, fainting, and unconsciousness. Chronic exposure to acetine can cause dizziness and sleepiness.	-
	Butanone (Methyl ethyl ketone)	Ketone	-	Butanone is a colorless fairly volatile liquid with a pleasant pungent odor. Flash point 20 °F. Vapors heavier than air. Does not react with water or many common materials. Stable in normal transportation.	Acute inhalation exposure to methyl ethyl ketone in humans results in irritation to the eyes, nose, and throat. Limited information is available on the chronic effects of butanone in humans.	-
	Acetaldehyde	VOCs	2.2E-06	Acetaldehyde is a colorless, flammable liquid used in the manufacture of acetic acid, perfumes, and flavors. It is also an intermediate in the metabolism of alcohol. It has a general narcotic action and also causes irritation of mucous membranes. Large doses may cause death from respiratory paralysis.	Acute exposure to acetaldehyde results in effects including irritation of the eyes, skin, and respiratory tract. Symptoms of chronic intoxication of acetaldehyde resemble those of alcoholism.	Group 2B
	Benzene	VOCs	6.0E-06	Benzene is a clear, colorless, highly flammable and volatile, liquid aromatic hydrocarbon with a gasoline-like odor. Benzene is found in crude oils and as a by-product of oil-refining processes. In industry benzene is used as a solvent, as a chemical intermediate, and is used in the synthesis of numerous chemicals	Acute inhalation exposure of humans to benzene may cause drowsiness, dizziness, headaches, as well as eye, skin, and respiratory tract irritation, and, at high levels, unconsciousness. Chronic inhalation exposure has caused various disorders in the blood, including reduced numbers of red blood cells and aplastic anemia, in occupational settings.	Group 1
	Carbon disulphide	VOCs	-	Carbon disulphide is a colorless, flammable, poisonous liquid, CS2. It is used as a solvent, and is a counterirritant and has local anesthetic properties but is not used as such. It is highly toxic with pronounced CNS, hematologic, and dermatologic effects.	Acute inhalation exposure of humans to carbon disulfide has caused changes in breathing and chest pains. Nausea, vomiting, dizziness, fatigue, headache, mood changes, lethargy, blurred vision, delirium, and convulsions have also been reported in humans acutely exposed by inhalation. Neurologic effects, including behavioral and neurophysiological changes, have been observed in chronic human and animal inhalation studies. Reproductive effects, such as decreased sperm count and menstrual disturbances, have been observed in humans exposed to carbon disulfide by inhalation.	-
	Carbon tetrachloride	VOCs	6.0E-06	Carbon Tetrachloride is a clear, colorless, volatile and very stable chlorinated hydrocarbon. Carbon Tetrachloride is used as a solvent for oils and fats, as a refrigerant and as a dry-cleaning agent. Inhalation of its vapors can depress central nervous system activity and cause degeneration of the liver and kidneys. Carbon Tetrachloride is reasonably anticipated to be a human carcinogen based on evidence of carcinogenicity in experimental animals.	Human symptoms of acute inhalation to carbon tetrachloride include headache, weakness, lethargy, nausea, and vomiting. Acute exposures to higher levels and chronic inhalation exposure to carbon tetrachloride produces liver and kidney damage in humans.	Group 2B
	Chlorobenzene	VOCs	_	Chlorobenzene is a colorless to clear, yellowish liquid with a sweet almond-like odor. Chlorobenzene is used primarily as a solvent, a degreasing agent, and a chemical intermediate.	Acute exposure to chlorobenzene produced narcosis, restlessness, tremors, and muscle spasms. Chronic exposure of humans to chlorobenzene affects the central nervous system (CNS). Signs of neurotoxicity in humans include numbness, cyanosis, hyperesthesia (increased sensation), and muscle spasms.	-

em	Chemical	Species	Carcinogenic Risk IUR (µg/m3) ⁻¹	Characteristics	Consequence	IARC Group
<u> </u>	Chemical	Opecies	ιοιτ (μg/ιιιο)	Ona acteristics	Consequence	IARC Group
				Chloreform is a colorloss valatile, liquid derivative of trichloremethans with an other like oder. Formerly used	The major effect from acute (short-term) inhalation exposure to chloroform is central nervous system depression. Chronic (long-term)	
				as an inhaled anesthetic during surgery, the primary use of chloroform today is in industry, where it is used	exposure to chloroform by inhalation in humans has resulted in effects on the liver, including hepatitis and jaundice, and central nervous	
5	Chloroform	VOCs	2.3E-05	as a solvent and in the production of the refrigerant Freon.	system effects, such as depression and irritability.	Group 2B
					The major toxic effects caused by acute formaldehyde exposure via inhalation are eye, nose, and throat irritation and effects on the nasal	
					cavity. Other effects seen from exposure to high levels of formaldehyde in humans are coughing, wheezing, chest pains, and bronchitis.	
6	Formaldehyde	VOCs	1.3E-05	Formaldehyde is a colorless, flammable gas at room temperature and has a strong odor. Exposure to formaldehyde may cause adverse health effects.	Chronic exposure to formaldehyde by inhalation in humans has been associated with respiratory symptoms and eye, nose, and throat irritation.	Group 1
				, ,		·
				N-hexane is a clear colorless liquids with a petroleum-like odor. Flash points -9 °F. Less dense than water	Acute inhalation exposure of humans to high levels of hexane causes mild central nervous system (CNS) effects, including dizziness, giddiness, slight nausea, and headache.	
				and insoluble in water. Vapors heavier than air. Used as a solvent, paint thinner, and chemical reaction	Chronic exposure to hexane in air is associated with polyneuropathy in humans, with numbness in the extremities, muscular weakness,	
	Hexane (or n-hexane)	VOCs	-	medium.	blurred vision, headache, and fatigue observed.	-
	Methanol	VOCs	-	A colorless, flammable liquid used in the manufacture offormaldehyde and acetic acid, in chemical synthesis, antifreeze, and as a solvent. Ingestion of methanol is toxic and may cause blindness.	Acute or chronic exposure of humans to methanol by inhalation or ingestion may result in blurred vision, headache, dizziness, and nausea	-
				,	, , , , , , , , , , , , , , , , , , , ,	
				Methyl chloride is a colorless gas with a faint sweet odor. Shipped as a liquid under its vapor pressure. A leak may either be liquid or vapor. Contact with the liquid may cause frostbite by evaporative cooling. Easily		
	Mother dealers of the			ignited. Vapors heavier than air. Can asphyxiate by the displacement of air. Under prolonged exposure to fire	Acute exposure to high concentrations of methyl chloride in humans has caused severe neurological effects. Methyl chloride has also	
	Methyl chloride (Chloromethane)	VOCs	-	or intense heat the containers may rupture violently and rocket. Used to make other chemicals and as a herbicide.	caused effects on the heart rate, blood pressure, liver, and kidneys in humans. Chronic exposure have shown liver, kidney, spleen, and central nervous system (CNS) effects.	Group 3
						·
				Methyl chloroform is a colorless liquid with a sweet, pleasant odor. May irritate skin, eyes and mucous	Symptoms of acute inhalation exposure include dizziness, nausea, vomiting, diarrhea, loss of consciousness, and decreased blood pressure	
	Methyl chloroform			membranes. In high concentrations the vapors may have a narcotic effect. Nonflammable, but may	in humans.	
	(1,1,1-Trichloroethane)	VOCs	-	decompose and emit toxic chloride fumes if exposed to high temperatures. Used as a solvent.	After chronic inhalation exposure to methyl chloroform, some liver damage was observed in mice and ventricular arrhythmias in humans.	Group 3
				Methylene Chloride is a clear, colorless, nonflammable, volatile liquid chlorinated hydrocarbon with a sweet,		
				pleasant smell and emits nightly toxic turnes of phosgene when heated to decomposition, methylene chloride is primarily used as a solvent in paint removers, but is also used in aerosol formulations, as a solvent in the	The acute ffects of methylene chloride inhalation in humans consist mainly of nervous system effects including decreased visual, auditory, and motor functions, but these effects are reversible once exposure ceases.	
	Methylene chloride	V00-	4.05.00	manufacture of pharmaceuticals, as a degreasing agent, in electronics manufacturing and as an ethane	The effects of chronic exposure to methylene chloride suggest that the central nervous system (CNS) is a potential target in humans and	0
	(Dichloromethane)	VOCs	1.0E-08	foam blowing agent.	animals.	Group 2A
					Acute exposure to styrene in humans results in mucous membrane and eye irritation, and gastrointestinal effects.	
				Styrene is a colorless, toxic liquid with a strong aromatic odor. It is used to make rubbers, polymers and	Chronic exposure to styrene in humans results in effects on the central nervous system (CNS), such as headache, fatigue, weakness, and	
	Styrene	VOCs	-	copolymers, and polystyrene plastics.	depression, CSN dysfunction, hearing loss, and peripheral neuropathy.	Group 2B
					Acute inhalation exposure to very high levels of 1,1,2,2-tetrachloroethane has resulted in effects on the liver and respiratory, central nervous,	
	1,1,2,2-				and gastrointestinal systems in humans. Chronic inhalation exposure to 1,1,2,2-tetrachloroethane in humans results in jaundice and an enlarged liver, headaches, tremors,	
	Tetrachloroethane	VOCs	5.8E-05	1,1,2,2-tetrachloroethane is a colorless to pale yellow liquid with a sweet odor. Sinks in water.	dizziness, numbness, and drowsiness.	Group 2B
					Effects resulting from courts high local inholation purposes of humans to take obligate the local include invitation of the upper receivation than	
				Tetrachloroethylene is a colorless, volatile, nonflammable, liquid, chlorinated hydrocarbon with an ether-	Effects resulting from acute high-level inhalation exposure of humans to tetrachloroethylene include irritation of the upper respiratory tract and eyes, kidney dysfunction, and neurological effects such as reversible mood and behavioral changes, impairment of coordination,	
	T-to-able as attended a			like odor that may emit toxic fumes of phosgene when exposed to sunlight or flames. Tetrachloroethylene is	dizziness, headache, sleepiness, and unconsciousness. The primary effects from chronic inhalation exposure are neurological, including	
	Tetrachloroethylene (Perchloroethylene)	VOCs	2.6E-07	mainly used as a cleaning solvent in dry cleaning and textile processing and in the manufacture of fluorocarbons.	impaired cognitive and motor neurobehavioral performance. The IARC has classified trichloroethylene as Group 2A.	Group 2A
					The central nervous system (CNS) is the primary target organ for toluene toxicity in humans for acute and chronic exposures. CNS	
				Toluene is a clear colorless liquid with a characteristic aromatic odor. Flash point 40 °F. Less dense than	dysfunction and narcosis have been frequently observed in humans acutely exposed to elevated airborne levels of toluene; symptoms	
				water (7.2 lb / gal) and insoluble in water. Hence floats on water. Vapors heavier than air. May be toxic by inhalation, ingestion or skin contact. Used in aviation and automotive fuels, as a solvent, and to make other	include fatigue, sleepiness, headaches, and nausea. CNS depression has been reported to occur in chronic abusers exposed to high levels of toluene. Chronic inhalation exposure of humans to toluene also causes irritation of the upper respiratory tract and eyes, sore throat.	
	Toluene	VOCs	-	chemicals.	dizziness, and headache.	Group 3
				1,2,4-trichlorobenzene is a colorless liquid or white solid with a sharp chlorobenzene odor. Melting point	Acute exposure of 1,2,4-Trichlorobenzene can irritate the nose and throat.	
	1,2,4-Trichlorobenzene	VOCs	-	16.95 °C (62.5 °F) .	Chronic expoure of 1,2,4-Trichlorobenzene can cause damage to liver and kidneys.	-
					Acute exposure to very high levels of 1,1,2,2-tetrachloroethane has caused severe liver destruction in humans. Respiratory and eye irritation,	
					dizziness, nausea, and vomiting have been noted in humans exposed to fumes at high levels in the workplace.	
				1,1,2-Trichloroethane is a colourless, sweet-smelling liquid that does not dissolve in water, but is soluble in	Chronic exposure of humans to high levels of 1,1,2,2-tetrachloroethane results in effects on the liver (jaundice and an enlarged liver), central and peripheral nervous system (headaches, tremors, dizziness, and drowsiness), and gastrointestinal effects (pain, nausea, vomiting, and	
	1,1,2-Trichloroethane	VOCs	1.6E-05	most organic solvents. It is an isomer of 1,1,1-trichloroethane.	loss of appetite).	Group 3
				Trichloroethylene is a synthetic, light sensitive, volatile, colorless, liquid that is miscible with many non-polar	Acute and chronic inhalation exposure to trichloroethylene can affect the human central nervous system (CNS), with symptoms such as dizziness, headaches, confusion, euphoria, facial numbness, and weakness. Liver, kidney, immunological, endocrine, and developmental	
	Trichloroethylene	VOCs	4.3E-07	organic solvents. Trichloroethylene is used mainly as a degreaser for metal parts.	effects.	Group 1

			Carcinogenic Risk			
tem	Chemical	Species	· · · ·	Characteristics	Consequence	IARC Group
				It is a colorless, sweet-smelling liquid or gas occurring naturally in petroleum, coal and wood tar, and is so	Acute inhalation exposure to mixed xylenes in humans results in irritation of the eyes, nose, and throat, gastrointestinal effects, eye irritation,	
29	Xylenes (isomers and mixture)	VOCs		named because it is found in crude wood spirit Xylenes are released into the atmosphere as fugitive emissions from industrial sources, from auto exhaust, and through volatilization from their use as solvents.	and neurological effects. Chronic inhalation exposure of humans to mixed xylenes results primarily in central nervous system (CNS) effects, such as headache, dizziness, fatique, tremors, and incoordination; respiratory, cardiovascular, and kidney effects.	Group 3
.5	mixture)	VOCS	<u> </u>	emissions from moustrial sources, from auto exhaust, and unough voidulization from their use as solvents.	Social as neadache, dizziness, fatigue, tiernois, and incoordination, respiratory, cardiovascular, and kidney effects.	Group 3
				Methyl mercaptan is a colorless gas with a smell like rotten cabbage. It is a natural substance found in the		
				blood, brain, and other tissues of people and animals. It is released from animal feces. It occurs naturally in	Acute health effects of Methyl mercaptan can irritate the nose and throat. Breathing Methyl mercaptan can irritate the lungs causing	
				certain foods, such as some nuts and cheese. Methyl mercaptan is released from decaying organic matter in	coughing and/or shortness of breath. Higher exposures can casue a build-up of fluid in the lungs, a medical emergency, with severe	
	Methyl mercaptan			marshes and is present in the natural gas of certain regions in the United States, in coal tar, and in some crude oils. It is manufactured for use in the plastics industry, in pesticides, and as a jet fuel additive. It is also	shortness of breath. Exposure to methyl mercaptan can casue headache, nausea, vomiting, dizziness, muscle weakness and loss of coordination. Higher levels can cause loss of consciousness and death.	
0	(Methanethiol)	VOCs	-	released as a decay product of wood in pulp mills.	No information is currently available for the chronic human health effect for methyl mercaptan.	-
				Ethyl mercaptan is a clear colorless low-boiling liquid (boiling point 97 °F) with an overpowering, garlic-		
	Ethyl margantan			like/skunk-like odor. Flash point -55 °F. Less dense than water and very slightly soluble in water. Vapors are	Unhalation of vapor equace munaular weakness, convulsions, respiratory paralysis. High concentrations may equac pulmonary irritation	
1	Ethyl mercaptan (Ethanethiol)	VOCs	_	heavier than air. Vapors may irritate nose and throat. May be toxic if swallowed, by inhalation or by contact. Added to natural gas as an odorant. Used as a stabilizer for adhesives.	Inhalation of vapor causes muscular weakness, convulsions, respiratory paralysis. High concentrations may cause pulmonary irritation. Liquid irritates eyes and skin. Inqestion causes nausea and irritation of mouth and stomach.	-
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				1,2-Dichloroethane is a clear, colorless, oily, synthetic, flammable liquid chlorinated hydrocarbon with a	Acute and chronic exposure to 1,2-Dichloroethane by inhalation has resulted in a variety of effects in humans, including effects on the central	
2	1,2-Dichloroethane	VOCs	2.6E-05	pleasant chloroform-like smell that emits toxic fumes of hydrochloric acid when heated to decomposition.	nervous system, liver, kidney, lung, and cardiovascular system	Group 2B
				Ethylbenzene is a clear colorless liquid with an aromatic odor. Flash point 59 °F. Less dense than water (at	Acute exposure to ethylbenzene in humans results in respiratory effects, such as throat irritation and chest constriction, irritation of the eyes, and neurological effects such as dizziness.	
				7.2 lb / gal) and insoluble in water. Hence floats on water. Vapors heavier than air. Used as a solvent and to	Chronic exposure to ethylbenzene by inhalation in humans has shown conflicting results regarding its effects on the blood. Animal studies	
3	Ethylbenzene	VOCs	2.5E-06	make other chemicals.	have reported effects on the blood, liver, and kidneys from chronic inhalation exposure to ethylbenzene.	Group 2B
					Acute exposure to a-Pinene can irritate the nose and throat causing coughing and wheezing. Exposure to a-Pinene can cause headache,	
					nausea and vomiting.	
ı	a-Pinene	VOCs		Alpha-pinene is a clear colorless liquid with a turpentine odor. Flash point 91 °F. Less dense than water and insoluble in water. Vapors are heavier than air. Used as a solvent.	Chronic exposure to a-Pinene can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath. a-Pinene may also damage the kidneys.	
•	a-rillelle	VOCS	-	insoluble in water. Vapors are neavier than air. Oseu as a solvent.	Shortness of breath, a-Finetie may also damage the kidneys.	-
				n-decane is a colorless liquid. Flash point 115 °F. Less dense than water and insoluble in water. Vapors	Contact with eyes may produce mild irritation. Contact with skin may cause defatting, redness, scaling, and hair loss. Ingestion may cause	
5	n-Decane	VOCs	-	heavier than air. In high concentrations its vapors may be narcotic. Used as a solvent and to make other chemicals.	diarrhea, slight central nervous system depression, difficulty in breathing and fatigue. Inhalation of high concentrations may cause rapid breathing, fatigue, headache, dizziness, and other CNS effects.	-
					Symptoms of exposure to this compound may include irritation and sensitization of the skin. It may also cause eye irritation and damage.	
					Ingestion of large doses may lead to albuminuria and hematuria. This type of compound irritates all tissues intensely and may cause	
					circulatory collapse. Ingestion of this type of compound may cause abdominal burning, nausea, vomiting, diarrhea, dysuria, hematuria,	
					unconsciousness, shallow respiration, and convulsions. Inhalation of this type of compound may cause dizziness, rapid and shallow breathing, tachycardia, bronchial irritation, unconsciousness and convulsions. Anuria, pulmonary edema and bronchia pneumonia may	
6	d-Limonene	VOCs	-	D-limonene is a clear colorless mobile liquid with a pleasant lemon-like odor.	complicate recovery after either type of exposure.	Group 3
				Terpenes (/ˈtɜːrpiːn/) are a large and diverse class of organic compounds, produced by a variety of plants,	The same of the state of the st	
,	Terpenes	VOCs	_	particularly conifers, though also by some insects such as termites or swallowtail butterflies, which emit terpenes from their osmeteria.	They are often strong-smelling. They may protect the plants that produce them by deterring herbivores and by attracting predators and parasites of herbivores. Many terpenes are aromatic hydrocarbons and thus may have had a protective function.	-
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				Dishlasharan is an arrain same at 1919 () 1 0011000 Till at 1919 ()		
	o-Dichlorobenzene (1,2-			o-Dichlorobenzene is an organic compound with the formula C6H4Cl2. This colourless liquid is poorly soluble in water but miscible with most organic solvents. It is a derivative of benzene, consisting of two		
3	Dichlorobenzene)	VOCs	-	adjacent chlorine centers.	The acute effects of o-Dichlorobenzene by inhalation are reported to be headache, nausea, vomiting, vertigo, malaise and unconsciousness.	Group 3
					Acute exposure to m- Dichlorobenzene can irritate the nose and throat causing coughing and wheezing but also headache, drowsiness,	
	m-Dichlorobenzene(1,3-			m- Dichlorobenzene is a colorless liquid that is insoluble to water in which is used to make herbicides,	nausea, vomiting diarrhea and abdominal cramps. It may also damage the red blood cells leading to low blood count (anemia).	
1	Dichlorobenzene)	VOCs		insecticides, medicine, and dyes.	No information is currently available for the chronic human health effect for m- Dichlorobenzene.	Group 3
					When 1,4-dichlorobenzene is heated to decomposition, toxic gases and vapors (such as hydrochloric acid and carbon monoxide) are	
				1 4-Dichlorobenzene is a synthetic, white crystalline solid that is practically insoluble in water and soluble in	released. The primary route of potential human exposure to this compound is inhalation. Acute inhalation exposure to 1,4-	
	p-Dichlorobenzene (1,4-			ether, chloroform,carbon disulfide, benzene, alcohol and acetone. It is used primarily as a space deodorant in products such as room deodorizers, urinal and toilet bowl blocks, and as an insecticide fumigant for moth	dichlorobenzene can result in coughing and breathing difficulties. Breathing high levels of this chemical can cause headaches, dizziness and liver damage. Contact with 1,4-dichlorobenzene can irritate the eyes, leading to burning and tearing. It is reasonably anticipated to be a	
)	Dichlorobenzene)	VOCs	1.1E-05	control.	human carcinogen.	Group 2B
				Naphthalene is a white, volatile, solid polycyclic hydrocarbon with a strong mothball odor. Naphthalene is		
				obtained from either coal tar or petroleum distillation and is primarily used to manufacture phthalic anhydride,	Acute exposure of humans to naphthalene by inhalation, ingestion, and dermal contact is associated with hemolytic anemia, damage to the	
	Naphthalene	PAHs	3.4E-05	but is also used in moth repellents.	liver, and neurological damage. Chronic exposure of workers and rodents to naphthalene would cause cataracts and damage to the retina.	Group 2B

Acenaphthylene is a colorless crystalline solid. Insoluble in water. Used in dye synthesis, insecticides. Acenaphthylene may cause headaches, fatigue exposure may irritate the eyes, nose and anorexia and anemia. Acenaphthylene may cause headaches, fatigue exposure to some polycyclic aromatic hydrocate in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acid smoke and irritating furnes when heated to decomposition. Derived from coal far and used to make dyes. Acenaphthene may cause headaches, fatigue, pharmaceuticals, insecticides, fungicides, and plastics. Fluorene PAHs Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate for chronic exposure, fluoren has been tested. Phenanthrene PAHs Phenanthrene PAHs Phenanthrene is a colorless monoclinic crystals with a faint aromatic odor. Solutions exhibit a blue dissiliable as to its potential to cause cancer. Anthracene, also called paranaphthalene or green oil, a solid polycyclic aromatic hydrocarbon. It is on the EPA's priority politurant list. It is ubsique on the environment as a product of incomplete combustion of fossit fuels. It has been indefined in surface and cigars, and in smoked foods and edible aquato cryanisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic semiconductor research. Anthracene Anthracene Anthracene is a colorless monoclinic order, smoke screens, scintillation counter crystals, and in organic hydrocarbon. It is on the EPA's priority politurant list. It is ubsique organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic hydrocarbon. It is on the little and the production of dyes, smoke screens, scintillation counter crystals, and in organic hydrocarbon. It is on the little and the	IARC Group
Berzo (a) Pyrene PAHs 8.7E-02 Berzo (a) Pyrene is a pale yellow, crystalline solid or powder waith a faint aromatic odor. In its pure form it is used as a laboratory reagent. It also forms as a gaseous by-product when certain carbon substances burn, and this chemical can increase these effects, such as coal tar chemicals, and is found in cigarette smoke. Acenaphthylene PAHs - Acenaphthylene is a coloriess crystalline solid. Insoluble in water. Used in dye synthesis, insecticides, and can be maintained to plassios. Acenaphthylene is a coloriess crystalline solid. Insoluble in water. Used in dye synthesis, insecticides, and and exposure may irritate the eyes, nose and ancrexia and anemia. Acenaphthylene is a colories crystalline solid. Insoluble in water. Used in dye synthesis, insecticides, and cancers and an emia. Acenaphthylene is a white needes. Melting point 93.6°C. Soluble in hot alcohol. Denser than water and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrist smoke and irritating furnes when heated to decomposition. Derived from coal tar and used to make dyes. Acenaphthene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate in coal and emia. Acenaphthene can irritate the diassifiable as to its potential to cause cancer. Anthracene PAHs - Phenanthrene is a colories monocinic crystals with a faint aromatic odor. Solutions exhibit a blue diassifiable as to its potential to cause cancer. Anthracene PAHs - Benanthrene is a colories monocinic crystals with a faint aromatic odor. Solutions exhibit a blue diassifiable as to its potential to cause cancer. Anthracene PAHs - Benanthrene is a colories monocinic crystals with a faint aromatic odor. Solutions exhibit a blue diassifiable as to its potential to cause cancer. Anthracene PAHs - senionductor research. Anthracene PAHs - senionductor research in model foods and edible aquatic organisms. It is primarily used as an intermediate in the production of	IARC Group
42 Benzo (a) Pyrene PAHs 8.7E-02 such as a laboratory reagent. It also forms as a gaseous by-product when certain carbon substances burn. Active exposure may irritate the eyes, nose and ancrexia and anernia. Acenaphthylene PAHs - Acenaphthylene is a colorless crystalline solid. Insoluble in water. Used in dye synthesis, insecticides, Acenaphthylene may cause headaches, fatigue, exposure to some polycyclic aromatic hydrocarto insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emits acrid smoke and ancrexia and anernia. Acenaphthene and accent and accent and used to make dyes, pharmaceuticals, insecticides, fungicides, and plastics. Fluorene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate Phenanthrene PAHs - Fluorene is white crystalline pl	
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Acenaphthylene PAHS - fungicides, and in the manufacture of plastics. Acenaphthylene is a white needles. Melting point 93.6 °C. Soluble in hot alcohol. Denser than water and insoluble in water. Hence sinks in water. May irritate skin and mucous membranes. Emils acrid smoke and irritating fumes when heated to decombistion. Derived from coal tar and used to make dyes, pharmaceuticals, insecticides, fungicides, and plastics. Acenaphthene PAHS - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate For chronic exposure, fluorene has been tested Phenanthrene PAHS - Fluorene is a colorless monoclinic crystals with a faint aromatic odor. Solutions exhibit a blue fluorescence. Anthracene, also called paranaphthalene or green oil, a solid polycyclic aromatic hydrocarbon. It is on the EPA's priority pollutant list. It is usuation in the single combustion of lossiff fuels. It has been identified in surface and drinking water, ambient air, exhaust emissions, smoke of cigarettes and cigars, and in smoked foods and edible aquatic organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic in coal tar products used in coating for pipes a to phenol) may include contact burns to the skir respiratory areas, in addition, NIOSH recomment.	d throat. Very high levels can cause headache, restlessness, lethargy, nausea, vomiting, e, and nausea with chronic exposre. Acenaphthylene may casue a skin allergy. Chronic
insoluble in water. Hence sinks in water, May irritate skin and mucous membranes. Emits acrid smoke and irritating furnes when heated to decomposition. Derived from coal tar and used to make dyes, pharmaceuticals, insecticides, fungicides, and plastics. 45 Fluorene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate For chronic exposure to some polycyclic aromatic hydrocard. 46 Phenanthrene PAHs - Fluorene is a colorless monoclinic crystals with a faint aromatic odor. Solutions exhibit a blue classifiable as to its potential to cause cancer. 47 Anthracene PAHs - Spriority pollutant list. It is ubiquitous in the environment as a product of incomplete combustion of tossil fuels. It has been identified in surface and drinking water, ambient air, exhaust emissions, smoke of cigarettes and cigars, and in smoked foods and edible aquatic organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic semiconductor research. 47 Anthracene PAHs - semiconductor research. 48 Anthracene PAHs - semiconductor research. 49 Anthracene PAHs - semiconductor research.	rbons (PAHs) may result in optic neuritis, lens opacities, and chorioretinitis.
Fluorene PAHs - Fluorene is white crystalline plates. It is used in resinous products, dyestuffs, and as a chemical intermediate For chronic exposure, fluornen has been tested Phenanthrene PAHs - Phenanthrene is a colorless monoclinic crystals with a faint aromatic odor. Solutions exhibit a blue dassifiable as to its potential to cause cancer. Anthracene, also called paranaphthalene or green oil, a solid polycyclic aromatic hydrocarbon (PAH) consisting of three benzene rings derived from coal-tar, is the simplest tricyclic aromatic hydrocarbon. It is on the EPA's priority pollutant list. It is ubiquitous in the environment as a product of incomplete combustion of fossil fuels. It has been identified in surface and drinking water, ambient air, exhaust emissions, smoke of cigarettes and cigars, and in smoked foods and edible aquatic organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic semiconductor research. Toxic by all routes (ie, inhalation, ingestion, den in coal tar products used in coatings for pipes a to phenol) may include contact burns to the significancy arrest. In addition, NIOSH recomme	d throat. Very high levels can cause headache, restlessness, lethargy, nausea, vomiting, , and nausea with chronic exposre. Acenaphthylene may casue a skin allergy. Chronic rbons (PAHs) may result in optic neuritis, lens opacities, and chorioretinitis. Group 3
46 Phenanthrene PAHs - fluorescence. classifiable as to its potential to cause cancer. Anthracene, also called paranaphthalene or green oil, a solid polycyclic aromatic hydrocarbon (PAH) consisting of three benzene rings derived from coal-tar, is the simplest tricyclic aromatic hydrocarbon. It is on the EPA's priority pollutant list. It is ubiquitous in the environment as a product of incomplete combustion of fossil fuels. It has been identified in surface and drinking water, ambient air, exhaust emissions, smoke of cigarettes and cigars, and in smoked foods and edible aquatic organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic semiconductor research. Anthracene PAHs - Toxic by all routes (ie, inhalation, ingestion, der in coal tar products used in coatings for pipes a to phenol) may include contact burns to the skir respiratory arrest. In addition, NIOSH recomme	in for acute health effects. d but further studies are requuired to determine its ability to cause cancer. Group 3
consisting of three benzene rings derived from coal-tar, is the simplest tricyclic aromatic hydrocarbon. It is on the EPA's priority pollutant list. It is ubiquitous in the environment as a product of incomplete combustion of fossil fuels. It has been identified in surface and drinking water, ambient air, exhaust emissions, smoke of cigarettes and cigars, and in smoked foods and edible aquatic organisms. It is primarily used as an intermediate in the production of dyes, smoke screens, scintillation counter crystals, and in organic 47 Anthracene PAHs - Mathracene PAHs - Semiconductor research. Toxic by all routes (ie, inhalation, ingestion, der in coal tar products used in coatings for pipes a to phenol) may include contact burns to the skir respiratory arrest. In addition, NIOSH recomme	ne nose and throat. While chronic human effects of Phenanthrene has been tested, it is not Group 3
Toxic by all routes (ie, inhalation, ingestion, der in coal tar products used in coatings for pipes a to phenol) may include contact burns to the skir respiratory arrest. In addition, NIOSH recomme	ntact with eyes causes irritation. Group 3
prevent skin contact. All such equipment and clean condition. Alsom workers should wash the exposed to heat and flame, it may be decomposed.	rmal contact), exposure to this pale yellow, crystalline substance may occur from its presence and storage tanks, in roofing materials, and asphalt. Effects from exposure (based on analogy in and eyes, nausea, tachycardia, cardiac arrhythmias, liver injury, pulmonary edema, and ends that fluoranthene be regulated as an occupational carcinogen. In activities and situations cocur, wear a self-contained breathing apparatus, amd personal protective clothing to elothing should be removed in change rooms at the end of the workday, and maintained in a noroughly at the end of the workday. While fluoranthene's fire hazard is only slight when nosed by heat, emitting acrid smoke and fumes. Spills of hot coal tar products may be covered or disposal by rotary kiln or fluidized bed forms of incineration.
Pyrene is a carcinogenic agent and is absorbed Workers exposed to 3 to 5 mg/m3 of pyrene exl acute toxicity of pure PAHs appears low when a exclusively via the gastrointestinal and respirato	d by the skin. It is a skin irritant, a suspected mutagen, and an equivocal tumor-causing agent. khibited some teratogenic effects. Pyrene is a polycyclic aromatic hydrocarbon (PAH). The administered orally or dermally to rats or mice. Human exposure to PAHs is almost ory tracts, and approximately 99 percent is ingested in the diet. Despite the high y be exposed through food, there is currently little information available to implicate diet-
49 Pyrene PAHs - Pyrene is a colorless solid, solid and solutions have a slight blue fluorescence. derived PAHs as the cause of serious health eff	
Benz a anthracene is a crystalline, aromatic hydrocarbon consisting of four fused benzene rings, produced by incomplete combustion of organic matter. Benz(a)anthracene is primarily found in gasoline and diesel exhaust, tobacco and cigarette smoke, coal tar and coal tar pitch, coal combustion emissions, charcoal-broiled foods, amino acids, fatty acids and carbohydrate pyrolysis products, wood and soot smoke, and creosote, asphalt and mineral oils. This substance is used only for research purposes. Benz(a)anthracene is PAHs 1.1E-04 Benz(a)anthracene is a crystalline, aromatic hydrocarbon consisting of four fused benzene rings, produced by incomplete combustion emissions, charcoal-broiled foods, amino acids, fatty acids and carbohydrate pyrolysis products, wood and soot smoke, and creosote, asphalt and mineral oils. This substance is used only for research purposes. Benz(a)anthracene is When heated to decomposition this compound	I emits acrid smoke and irritating fumes. Group 2B
Chrysenes is an aromatic hydrocarbon in coal tar, allied to naphthalene and anthracene. It is a white	
51 Chrysene PAHs 1.1E-05 crystalline substance, C18H12, of strong blue fluorescence, but generally colored yellow by impurities. Toxic as chronic and actue hazard.	Group 2B
Benzo b fluoranthene is a colorless, aromatic hydrocarbon consisting of five fused rings and formed by the incomplete burning of organic matter. Benzo(b)fluoranthene is primarily found in gasoline exhaust, tobacco and cigarette smoke, coal tar, soot, amino acids and fatty acid pyrolysis products. This substance is used only for research purposes. Benzo(b)fluoranthene is reasonably anticipated to be a human carcinogen. When heated to decomposition, this compound	d emits acrid smoke and irritating fumes Group 2B

			Carcinogenic Risk			
Item	Chemical	Species	IUR (μg/m3) ⁻¹	Characteristics	Consequence	IARC Group
53	Benzo(k)fluoranthene	PAHs		Benzo k fluoranthene is a pale yellow aromatic hydrocarbon consisting of five fused rings and is produced by the incomplete combustion of organic matter. Benzo(k)fluoranthene is primarily found in gasoline exhaust, cigarette smoke, coal tar, coal and oil combustion emissions, lubricating oils, used motor oils and crude oils. This substance is used only for research purposes. Benzo(k)fluoranthene is reasonably anticipated to be a human carcinogen.	hen heated to decomposition this compound emits acrid smoke and irritating fumes.	Group 2B
54	Indeno (1,2,3-cd)pyrene	PAHs		Indeno 1 2 3-cd pyrene is a yellowish, aromatic hydrocarbon consisting of six fused rings and produced by the incomplete combustion of organic matter. Indene[1,2,3-cd]pyrene is primarily found in certain foods, gasoline and diesel exhaust, cigarette smoke, coal tar and coal tar pitch, soot and petroleum asphalt. This substance is used only for research purposes. Indene[1,2,3-cd]pyrene is reasonably anticipated to be a human carcinogen.	Toxic as chronic and actue hazard.	Group 2B
55	Dibenz(a,h)anthracene	PAHs		Dibenzo a h Anthracene is a crystalline, carcinogenic aromatic hydrocarbon consisting of five fused benzene rings, produced by the incomplete combustion of organic matter. Dibenzo(a,h)anthracene is primarily found in gasoline exhaust, tobacco smoke, coal tar, soot and certain food products, especially smoked and barbecued foods. This substance is used only for research purposes to induce tumorigenesis. Dibenzo(a,h)anthracene is a mutagen and is reasonably anticipated to be a human carcinogen.	This compound is harmful if swallowed or inhaled. It may cause irritation. When heated to decomposition it emits acrid smoke, irritating fumes and toxic fumes of carbon monoxide and carbon dioxide.	Group 2A
56	Benzo(g,h,i)perylene	PAHs		Benzo[ghi]perylene is a colorless to white crystalline solid. Water insoluble. Large, pale yellow-green plates (recrystallized from xylene)	Inhalation of material may be harmful. Contact may cause burns to skin and eyes. Inhalation of Asbestos dust may have a damaging effect on the lungs. Fire may produce irritating, corrosive and/or toxic gases. Some liquids produce vapors that may cause dizziness or suffocation Runoff from fire control may cause pollution.	Group 3

Notes:

Group 1 Carcinogenic to humans

Group 2A Probably carcinogenic to humans
Group 2B Possibly carcinogenic to humans

Group 3 Not classifiable as to its carcinogenicity to humans

Group 4 Probably not carcinogenic to humans